



R20 Regulation

Subject code:3B2AJ

**TKR COLLEGE OF ENGINEERING AND TECHNOLOGY**

(Autonomous, Accredited by NAAC with 'A+' Grade)

**B.Tech II Semester Supplementary Examinations, July 2025**

**ENGINEERING CHEMISTRY**

(ME)

Maximum Marks: 70

Date: 11.07.2025

Duration: 3 hours

- Note:
1. This question paper contains two parts A and B.
  2. Part A is compulsory which carries 20 marks. Answer all questions in Part A.
  3. Part B consists of 5 Units. Answer any one full question from each unit.
  4. Each question carries 10 marks and may have a, b, c, d as sub questions.

**Part-A**

| All the following questions carry equal marks (10X2M=20 Marks) |  | Marks | CO | BTL |
|--|--|-------|----|-----|
| 1  | What is the magnetic nature of 'N <sub>2</sub> ' molecule?               | 2M    | 1  | L1  |
| 2  | What is meant by Doping in case of Semiconductors?                       | 2M    | 1  | L1  |
| 3  | What is meant by Hardness of water?                                      | 2M    | 2  | L1  |
| 4  | What is reverse osmosis?   | 2M    | 2  | L1  |
| 5  | Write the applications of Lithium-ion battery?                           | 2M    | 3  | L1  |
| 6  | What is electro chemical series? Write any two applications?             | 2M    | 3  | L1  |
| 7  | Explain the mechanism of pitting corrosion with a suitable illustration. | 2M    | 4  | L1  |
| 8  | Define wet corrosion?  | 2M    | 4  | L1  |
| 9  | What is the difference between Gross and net calorific value?            | 2M    | 5  | L1  |
| 10   | Define proximate analysis?   | 2M    | 5  | L1  |

**Part-B**

| Answer All the following questions. (5X10M=50Marks) |   | Marks    | CO | BTL |
|---|---|----------|----|-----|
| 11  | a) Explain the postulates of Molecular Orbital theory.<br>b) Write the Energy level diagrams of O <sub>2</sub> ?                  | 5M<br>5M | 1  | L2  |
| OR  |   |          |    |     |
| 12  | a) Explain the salient features of Crystal field theory.<br>b) Explain the crystal field splitting in octahedral metal complexes. | 5M<br>5M | 1  | L2  |
| 13  | What is the principle of EDTA titration? How the permanent hardness of water is determined using EDTA method.                     | 10M      | 2  | L2  |
| OR  |   |          |    |     |
| 14  | a) Explain the treatment of potable water.<br>b) Explain in detail about boiler corrosion.  | 5M<br>5M | 2  | L2  |
| 15  | a) Derive Nernst equation for the potential of a single electrode?<br>b) Explain glass membrane electrode.                        | 5M<br>5M | 3  | L2  |
| OR  |   |          |    |     |
| 16  | Explain the construction and working of Lead -acid storage battery and  | 10M      | 3  | L2  |

|    |   |          |   |    |
|----|---|----------|---|----|
|    | write its applications.   |          |   |    |
| 17 | a) Write a note on cathodic protection.<br>b) Explain the factors effecting rate of corrosion.                                  | 5M<br>5M | 4 | L2 |
|    | OR  |          |   |    |
| 18 | Explain the mechanism involved in wet corrosion.  | 10M      | 4 | L2 |
| 19 | a) What are the characteristics of a Good fuel?<br>b) Discuss Fischer – Tropsch method for the preparation of synthetic petrol. | 5M<br>5M | 5 | L2 |
|    | OR  |          |   |    |
| 20 | a) Explain about refining of petroleum by fractional distillation method.<br>b) Write a note on i)LPG ii) CNG                   | 5M<br>5M | 5 | L2 |