



TKR Group of Institutions

**B.Tech. I Semester Supplementary Examinations, January 2025**  
**ENGINEERING CHEMISTRY**  
( Common to ECE & CSE)

**Maximum Marks: 70**

**Date: 27.01.2025**

**Duration: 3 hours**

- Note:**
1. This question paper contains two parts A and B.
  2. Part A is compulsory which carries 20 marks. Answer all questions in Part A.
  3. Part B consists of 5 Units. Answer any one full question from each unit.
  4. Each question carries 10 marks and may have a, b, c, d as sub questions.

**Part-A**

All the following questions carry equal marks 10x20=20 Marks

1. Write the electron level (in term of  $O_2$ )
2. Write the geometrical nature of  $N_2$  molecule.
3. What is reverse osmosis?
4. Write any 4 applications of drinking water?
5. What is the significance of electro chemical series?
6. What are the disadvantages of corrosion?
7. Define E – Z factors? Give two examples.
8. What is Markownikoff rule.
9. What is chemical shift.
10. What is shielding and deshielding in NMR?

**Part-B**

Answer All the following questions. (10M X 5=50Marks)

11. (a) Explain the salient features of MOT theory. [5M]  
b. Explain bond order and magnetic properties of  $N_2$ . 5M  
OR
12. (a) Explain crystal field splitting in strong field square planar complex. [5M]  
(b) Write note on LCAG. [5M]
13. (a) Define hardness of water? How can we remove hardness of water by using ion exchange process. [5M]  
(b) A sample of water contains  $Mg^{2+} = 18 \text{ Mg/L}$ ,  $Ca^{2+} = 30 \text{ Mg/L}$ ,  $CO_2 = 11 \text{ Mg/L}$ ,  $HCO_3 = 122 \text{ mg/L}$ , then calculate temporary & permanent hardness of water in Degree Clark & Degree French. 5M  
OR
14. Explain the steps involved in potable water treatment. [10M]
15. (a) How do you determine the pH of an unknown solution by using Glass electrode? [5M]  
b. Write mechanism of galvanic corrosion. 5M  
OR

16	(a) What is battery ? Explain lead – acid storage cell. [5M] (b) Write a notes on cathodic protection of corrosion control method. [5M]
17	(a) What are SN <sup>1</sup> SN <sup>2</sup> reactions? Give examples. [5M] (b) Explain the different conformations of n – butane. [5M]
	OR
18	What are Addition reactions? Explain the possible mechanism with an example. [10M]
19	(a) Explain about various electronic transitions in UV spectroscopy? [5M] (b) Write a note on magnetic resonance imaging (MRI) [5M]
	OR
20	(a) Write a note on chemical shift. [5M] (b) Explain the types of vibrations in IR spectroscopy. [5M]